Scientific Facts on Phthalate
Di-isodecyl & Di-isononyl phthalates

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This Digest is a faithful summary of two leading scientific consensus reports
produced in 2003 by the European Chemicals Bureau (ECB):
“Summary Risk Assessment Report (RAR 041) on Di-isodecyl Phthalate (DIDP), 2003” and “Summary Risk
Assessment Report (RAR 046) on Di-isononyl Phthalate (DINP), 2003”

The full Digest is available at: https://www.greenfacts.org/en/dinp-didp/

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languages as questions and answers, in a copyrighted user-friendly Three-Level Structure of increasing
detail:

- Each question is answered in Level 1 with a short summary.
- These answers are developed in more detail in Level 2.
- Level 3 consists of the Source document, the internationally recognised scientific consensus
report which is faithfully summarised in Level 2 and further in Level 1.

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0. Introduction: What are phthalates?

Phthalates are widely used as additives in a range of plastics and other materials that are found in many consumer products. They make plastics, such as PVC, soft and flexible. They are not chemically bound to plastics, so they can be released from consumer products into the environment. There is public concern about phthalates because of their widespread use and occurrence in the environment as well as their potential effects on human health.

There is a wide range of different phthalates, which each have specific properties, uses, and health effects. In the European Union, five of the most widely used phthalates have been reviewed by the European Chemicals Bureau (DEHP, DBP, DINP, DIDP, and BBP).

By 2004, EU Risk Assessment Reports had been published for three of these phthalates: DIDP, DINP, and DBP; assessments which have been summarised by GreenFacts. Because of the strong similarities between the first two phthalates, they are described together.

1. What are the properties of DIDP and DINP?

DIDP and DINP are acronyms that each refer to certain mixtures of phthalates.

They have a common core structure with two long chains of molecules attached to the core. Most of these chains contain 10 carbon atoms in the case of DIDP, and 9 carbon atoms in the case of DINP. The arrangement of the molecules in the chains can differ and this is why DIDP and DINP are mixtures of closely-related substances.

DIDP and DINP are sticky, oily liquids. They are soluble in fat and not very soluble in water.

2. How are DIDP and DINP used?

In 1994, over 200 000 tonnes of DIDP and 107 000 tonnes of DINP were produced annually in the European Union and production is expected to increase further.

They are used mainly as a plasticiser in PVC that is used to make film, sheeting, coated products, flooring, roofing, wall coverings, hoses, tubing, wires, cables, injection moulded shoe soles, car undercoating and sealants.

They are also used in vinyl resins other than PVC, in cellulose ester plastics as well as in adhesives, paints and printing inks.

3. Can DIDP and DINP affect the environment?

Although DIDP and DINP are mixtures and each component can behave slightly differently in the environment, an overall picture can be drawn.

3.1 DIDP and DINP present in the environment can remain for a long time in soil and sediment. They can also concentrate in organisms living in water.
3.2 Most of the DIDP and DINP released into the general environment results from the use and disposal of PVC products containing them. DIDP and DINP can go into waste water, surface water and into air.

3.3 Local concentrations in the environment are particularly high around industrial sites where DIDP and DINP are either produced or used to make plastics and other products. At those sites concentrations of DIDP and DINP in water, sediment and soil are highest.

3.4 DIDP and DINP do not appear to have adverse effects on organisms in the environment. They are not toxic to microbes, plants or animals.

3.5 Despite the presence of DIDP and DINP in the environment, it is concluded that:
   • the levels found, they do not present a risk to the environment,
   • that no further testing is needed, and
   • that no further risk reduction measures need to be taken.

4. How can humans be exposed to DIDP and DINP?

Exposure of humans may occur because DIDP and DINP are present in the general environment, in the workplace and in consumer products.

4.1 The highest exposures can occur in certain workplaces where DIDP, DINP, or products that contain them are produced or used.

4.2 Exposure of the general public is about 50 times lower than workplace exposure. For infants, the level of exposure varies depending on whether they have contact with plastic toys and baby care equipment containing DIDP or DINP. If they have contact with such toys their exposure can be much greater.

5. What health effects can DIDP and DINP cause in laboratory animals?

DIDP and DINP are absorbed well when swallowed or breathed in, but are poorly absorbed through the skin.

In laboratory animals the main effects of DIDP and DINP are on the liver and include cancer. Humans are thought to be much less sensitive to these liver effects and the cancer is caused by a mechanism which is only seen in rodents and not in humans. DIDP and DINP also affect survival of rat offspring.

DIDP does not appear to interfere with hormones and reproduction, but DINP may have some impact on male hormones.

6. Do DIDP and DINP pose risks to human health?

6.1 When human exposures are compared with the lowest amounts needed to cause effects in laboratory animals, the margin of safety is sufficient for workers not to be considered at risk.

6.2 Exposure of the general public is lower than that of workers, thus adults and children above the age of three are not considered to be at risk.
In the worst case scenario for newborns and infants, which assumes that all the phthalates in toys and baby equipment are DIDP, the margin of safety is not sufficiently protective. This is because infants can absorb phthalates when putting toys into their mouth. If toys are excluded from exposure estimates, then infant exposures are not of concern.

For DINP, the margins of safety are considered sufficient, even for infants exposed to toys containing DINP.

7. Is further research needed?

The European Union Risk Assessment Report on DIDP and DINP (the source of the present summary) concluded that there is no need for further information or testing, or for further risk reduction measures beyond those already taken, except to limit risks should DIDP be used in toys.

The Scientific Committee on Toxicity, Ecotoxicity and the Environment has commented on the European Union Risk Assessment Report on DINP. It did not agree with several of the conclusions about environmental effects and considers that more research is necessary.

8. Conclusions

Phthalates have played an important role in the creation of plastics and other materials that have many versatile uses in industry, in medicine and in consumer products.

In view of more recent research, raising concerns about possible environmental and health effects, the risks of exposure to phthalates are being kept under close review by national and international bodies.

The most recent EU reviews on DIDP, DINP and DBP conclude that:
- more research may be necessary on the environmental effects of DIDP and DINP;
- DIDP in toys may pose a risk;
- in some workplaces, exposure to DBP should be reduced;
- release of DBP into the air from some workplaces should be reduced.

Other phthalates are currently being assessed by the European Chemicals Bureau.

GreenFacts comment:

Other EU reviews are currently underway and the information will be posted on this website as soon as it becomes available:
- **DEHP (Di-Ethyl-Hexyl-Phthalate) the most commonly used phthalate**
- **the use of phthalates in food packaging materials**
Annex

Annex 1:
Approximation of the relative importance of the consumption of four of the main phthalates in the European Union in the 1990s